

Mesomerization of S_4 -Symmetric Tetrahedral Chelate Complex $[\text{In}_4(\text{L}^3)_4]$: First-Time Monitored by Temperature-Dependent ^1H NMR Spectroscopy^[‡]

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Dedicated to Professor Paul von Ragué Schleyer on the occasion of his 80th birthday

Keywords: Bailar twist / NMR spectroscopy / Density functional calculations / Indium / N,O ligands / Supramolecular chemistry

VT ^1H NMR spectroscopy proved that a non-dissociative and reversible mesomerization process links the tetranuclear indium(III) complexes *meso*-($\Delta, \Delta, \Delta, \Delta$)-(P,P,M,M)-**3** and *meso*-($\Lambda, \Lambda, \Lambda, \Lambda$)-(M,M,P,P)-**3'**. During this process four tandem Bailar twists, resulting in the (Δ)/(Λ) isomerization at the indium

centers, and the (P)/(M) inversion of the four coordinating face-centered, helical ligands (L^3)³⁻ are involved. In addition, gas-phase DFT calculations (B3LYP/LANL2DZp) revealed a C_1 -symmetric transition state (+21.9 kcal mol⁻¹) for the mesomerization mechanism which connects **3** and **3'**.

Introduction

Metal-assisted self-assembly^[2] is a very convenient method for the synthesis of supramolecular aggregates. NMR spectroscopy was shown to be an effective tool for the investigation of these self-assembled coordination compounds and for the study of their dynamic behavior in solution.^[3,4] In this context, we reported on variable-temperature (VT) ^1H NMR spectroscopy studies revealing tetrahedral, racemic, homochiral ($\Delta, \Delta, \Delta, \Delta$)-[(NH₄)₄∩{Mg₄(L¹)₆}] ⇌ ($\Lambda, \Lambda, \Lambda, \Lambda$)-[(NH₄)₄∩{Mg₄(L¹)₆}] (NH₄-**1**) to enantioimerize.^[4a] This phenomenon can be explained by simultaneous Bailar twists associated with (Δ)/(Λ) isomerization at the four octahedrally coordinated magnesium centers synchronized with sterically unhindered (R)/(S) atropisomerization processes around the C–C single bonds of the six edge-bridging C_2 -symmetric ligands (L¹)²⁻. Supplementary support for this phenomenon is presented by additional studies of ($\Delta, \Delta, \Delta, \Delta$)-[(EtNH₃)₄∩{Mg₄(L¹)₆}] ⇌ ($\Lambda, \Lambda, \Lambda, \Lambda$)-[(EtNH₃)₄∩{Mg₄(L¹)₆}] (EtNH₃-**1**), where the diastereotopic methylene protons of the ethylammonium counterions of EtNH₃-**1** display similar temperature-de-

pendent coalescence as the ligand vinyl ether methylene protons (Figure 1). On the contrary, racemic ($\Delta, \Delta, \Delta, \Delta$)/($\Lambda, \Lambda, \Lambda, \Lambda$)-[(NH₄)₄∩{Mg₄(L²)₆}] (NH₄-**2**) showed to be kinetically and dynamically stable on the ^1H NMR timescale. Owing to steric hindrance by the extra CO₂Et groups, rotation around the central C–C bond in (L²)²⁻ is blocked, which prevents NH₄-**2** from enantiomerization (Figure 1).^[4a]

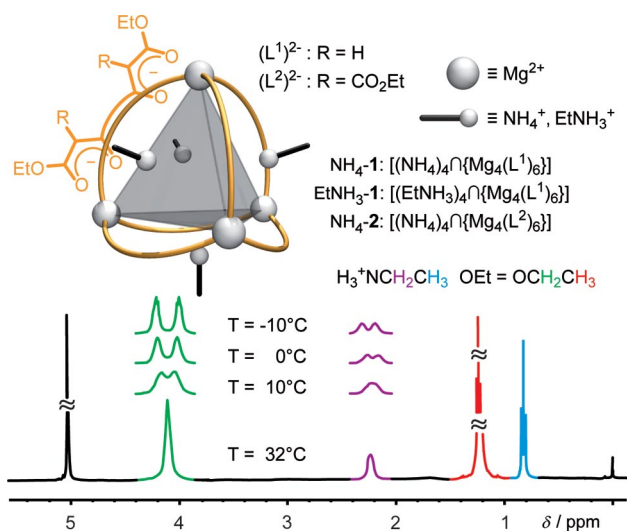


Figure 1. Top: Schematic representation of NH₄-**1**, EtNH₃-**1**, and NH₄-**2**. Bottom: VT ^1H NMR spectrum of EtNH₃-**1**.

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Results and Discussion

These VT ^1H NMR studies on the enantiomerization of $\text{NH}_4\text{-1}$, and $\text{EtNH}_3\text{-1}$ prompted us to also study the potential dynamic behavior of tetrahedral *meso*-($\Delta,\Delta,\Delta,\Delta$)(*P,P,M,M*)- $[\text{In}_4(\text{L}^3)_4]$ (**3**) and homochiral racemic ($\Delta,\Delta,\Delta,\Delta$)($\Lambda,\Lambda,\Lambda,\Lambda$)- $[\text{In}_4(\text{H}^{\text{N}}\text{L}^3)_4](\text{ClO}_4)_4$ (**4**) (Figure 2).

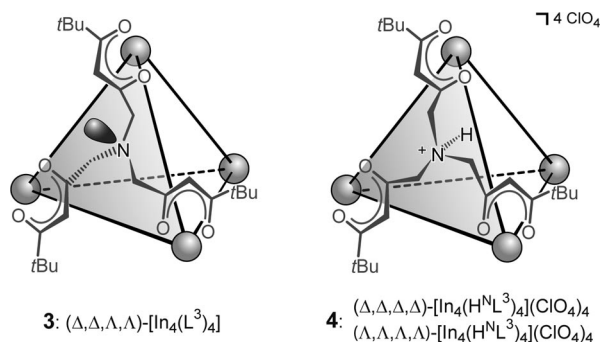


Figure 2. Schematic presentation of *meso*-($\Delta,\Delta,\Delta,\Delta$)(*P,P,M,M*)- $[\text{In}_4(\text{L}^3)_4]$ (**3**) and ($\Delta,\Delta,\Delta,\Delta$)($\Lambda,\Lambda,\Lambda,\Lambda$)- $[\text{In}_4(\text{H}^{\text{N}}\text{L}^3)_4](\text{ClO}_4)_4$ (**4**).

Based on an X-ray structural analysis, complex **3**^[5] is composed of four indium ions, located at the apices of a tetrahedron linked by four ligands (L^3)³⁻ across the tetrahedral faces. The distorted octahedrally coordinated indium ions have alternately (Δ) or (Λ) configuration. The C_3 symmetry of the ligands (L^3)³⁻ is broken during complexation to the indium ions. However, despite the desymmetrized C_1 -symmetric ligands (L^3)³⁻, **3** is intrinsically achiral. This is due to the (*P*)/(*M*) helicity of the ligands (L^3)³⁻ resulting in an overall S_4 molecule symmetry of *meso*-($\Delta,\Delta,\Delta,\Delta$)(*P,P,M,M*)- $[\text{In}_4(\text{L}^3)_4]$ (**3**) (Figure 3).^[6]

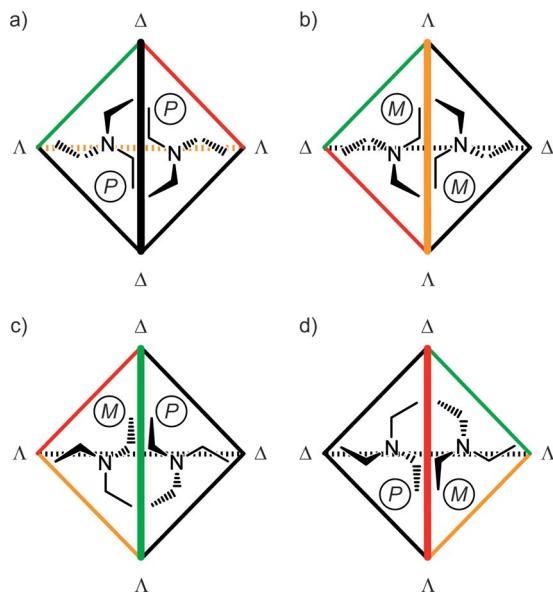


Figure 3. Schematic presentation of the (*P*)/(*M*) helicity of the C_1 -symmetric ligands (L^3)³⁻ of S_4 -symmetric **3**. View towards the edges: (a) (Δ,Δ) black, (b) (Λ,Λ) ocher, (c) (Δ,Λ) green, (d) (Λ,Δ) red.

In total agreement with the X-ray data,^[5] the ^1H NMR spectrum of **3** in $[\text{D}_8]\text{toluene}$ at 20°C displays two sets of three signals for the olefinic protons and *t*Bu groups ($\delta = 6.42, 6.33, 5.26$ ppm and $\delta = 1.32, 1.18, 1.00$ ppm, respectively). The diastereotopic CH_2 protons appear as three simple, but different, AB systems [$\delta = 4.30$ and 3.35 ppm, $|^2J_{\text{H,H}}| = 19.3$ Hz; $\delta = 3.73$ and 3.19 ppm, $|^2J_{\text{H,H}}| = 14.4$ Hz; $\delta = 3.09$ and 2.66 ppm, $|^2J_{\text{H,H}}| = 14.1$ Hz]. This proves **3** to be kinetically stable on the NMR timescale. Most interestingly, and according to what we expected, the ^1H NMR spectrum of *meso*-($\Delta,\Delta,\Delta,\Delta$)(*P,P,M,M*)- $[\text{In}_4(\text{L}^3)_4]$ (**3**) revealed temperature dependence. In the range of 20 – 105°C the signals of the olefinic protons (blue), the *t*Bu groups (red), and of the diastereotopic CH_2 protons (green) become broader and finally coalesce (Figure 4). The unique dynamic temperature-dependent ^1H NMR spectroscopic behavior of *meso*-**3** can be explained by an unprecedented mesomerization of the identical twins ($\Delta,\Delta,\Delta,\Delta$)(*P,P,M,M*)-**3** $\rightleftharpoons$ ($\Lambda,\Lambda,\Delta,\Delta$)(*M,M,P,P*)-**3'** (Figure 4). This mesomerization process is the first of this type and requires *four* tandem Bailar twists, which result in the inversion of the chirality at the indium centers together with the (*P*)/(*M*) inversion of the *four* coordinated C_1 -symmetric helical ligands (L^3)³⁻.^[7] It is worth to note, that **3** and **3'** are identical and would only be distinguishable in the case of two pairs of different metal ions. A prerequisite for the performance of the Bailar twists in **3** is the sufficiently flexible $[\text{In}_4(\text{L}^3)_4]$ scaffold. This is guaranteed, since the coordinated and thus helical C_1 -symmetric ligands (L^3)³⁻ allow simultaneous ste-

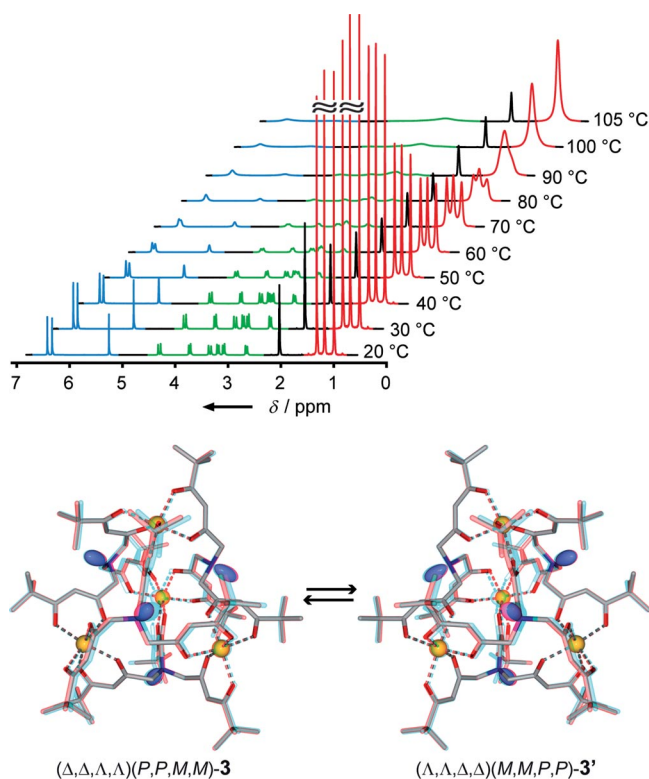


Figure 4. Top: VT ^1H NMR spectra of *meso*-**3**. Bottom: 3D presentation (red/blue) of the mesomerization ($\Delta,\Delta,\Delta,\Delta$)(*P,P,M,M*)-**3** $\rightleftharpoons$ ($\Lambda,\Lambda,\Delta,\Delta$)(*M,M,P,P*)-**3'**.

rically unhindered back and forth twists around the single bonds. Furthermore, this dynamic process goes along with concurrent helix (*P*)/(*M*) isomerization of the face-centered ligands and thus leads to the enantiotopization of the olefinic protons, *t*Bu groups, and diastereotopic CH_2 protons, as monitored by ^1H NMR spectroscopy. This mesomerization process is reversible and occurs non-dissociative without the formation of diastereomers. The high-temperature ^1H NMR spectra (Figure 4) indicate time-averaged pseudo-*T* molecular symmetry for **3** due to rapid, non-dissociative mesomerization ($\mathbf{3} \rightleftharpoons \mathbf{3}'$). As explicitly shown for the *t*Bu protons of **3**, coalescence is observed at 95 °C (400 MHz). The activation parameters for the molecular interconversion have been derived from line-shape analysis: $\Delta H^\ddagger = 11.8 \text{ kcal mol}^{-1}$, $\Delta S^\ddagger = -16 \text{ cal K}^{-1}$, $\Delta G^\ddagger_{368} = 17.7 \text{ kcal mol}^{-1}$ (see the Supporting Information).

Density functional calculations at B3LYP/LANL2DZp^[8–11] on the model complex $[\text{In}_4(\text{L}^*)_4]$ ^[12] suggest, that the mesomerization of $[\text{In}_4(\text{L}^*)_4]$ starts by (*P*)/(*M*) isomerization of a C_1 -symmetric helical ligand $(\text{L}^*)^{3-}$ at the nitrogen center, initiating the (Δ)/(Λ) isomerization at the ligated indium(III) centers. This leads to the subsequent (*P*)/(*M*) isomerization of the other three helical ligands $(\text{L}^*)^{3-}$ and the Bailar twists at the remaining metal center, resulting in the mesomerization of the whole complex. This corresponds to an activation barrier of $21.9 \text{ kcal mol}^{-1}$. The imaginary mode indicates that the mesomerization proceeds in a concerted, yet asynchronous fashion (Figure 5).

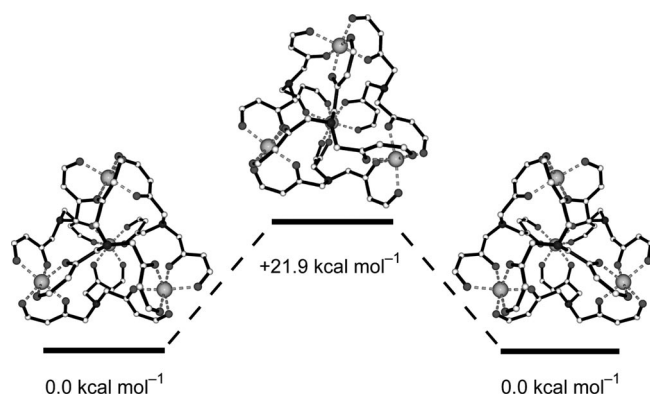


Figure 5. Energetic profile, illustrating the mesomerization of $[\text{In}_4(\text{L}^*)_4]$ in a concerted, yet asynchronous fashion. In gray, C white, O gray (small), N black.

Contrary to *meso*-**3**, the solid-state structure analysis of $[\text{In}_4(\text{H}^{\text{NL}}\text{L}^3)_4](\text{ClO}_4)_4$ (**4**) shows that the tetracation $(\mathbf{4})^{4+}$ is present in the solid state as a racemic mixture of homochiral $(\Delta, \Delta, \Delta, \Delta)/(\Lambda, \Lambda, \Lambda, \Lambda)$ -*fac* stereoisomers with the protons at the nitrogen atom directed towards the center of the tetrahedron.^[5] The ^1H NMR spectrum of $[\text{In}_4(\text{H}^{\text{NL}}\text{L}^3)_4]^{4+}$ (**4**)⁴⁺ in $[\text{D}_6]\text{dmsO}$ at 23 °C reveals one set of signals for the four equivalent ligands, indicating *T* molecule symmetry. However, owing to the fixation of the protons at the nitrogen atom of the ligands $(\text{H}^{\text{NL}}\text{L}^3)^{2-}$, enantioimerization of both enantiomers of racemic $(\Delta, \Delta, \Delta, \Delta)/(\Lambda, \Lambda, \Lambda, \Lambda)$ - $[\text{In}_4(\text{H}^{\text{NL}}\text{L}^3)_4](\text{ClO}_4)_4$ (**4**) is blocked, which is documented by

the temperature independence of the ^1H NMR spectra of **4** (see the Supporting Information).

Conclusions

We have shown that the complexes *meso*-($\Delta, \Delta, \Delta, \Delta$)-(*P, P, M, M*)-**3** and *meso*-($\Lambda, \Lambda, \Delta, \Delta$)-(*M, M, P, P*)-**3'** are identical and are correlated with each other by four tandem Bailar twists, which results in the (Δ)/(Λ) isomerization at the indium centers and the (*P*)/(*M*) isomerization of the four coordinating helical ligands $(\text{L}^3)^{3-}$. This mesomerization process is non-dissociative and reversible as was demonstrated by VT ^1H NMR spectroscopy. The mechanism connecting **3** and **3'** was elucidated by DFT calculations revealing a C_1 -symmetric transition state. However, owing to steric fixation, (*P*)/(*M*) isomerization of the ligands $(\text{H}^{\text{NL}}\text{L}^3)^{2-}$ of racemic $(\Delta, \Delta, \Delta, \Delta)/(\Lambda, \Lambda, \Lambda, \Lambda)$ - $[\text{In}_4(\text{H}^{\text{NL}}\text{L}^3)_4](\text{ClO}_4)_4$ (**4**) is blocked and thereby the enantioimerization of **4**, as shown by ^1H NMR spectroscopy.

Experimental Section

General Methods: All reagents and solvents employed were commercially available high-grade purity materials (Fluka, Aldrich), which were used as supplied without further purification. Synthesis and further spectroscopic data of **3** and **4** were reported previously.^[5] Temperature-dependent ^1H NMR spectra of **3** and **4** were recorded with a JEOL EX400 spectrometer (400 MHz) in indicated solutions. The residual solvent signal ($\delta = 2.03$ and 2.49 ppm for $[\text{D}_8]\text{toluene}$ and $[\text{D}_6]\text{dmsO}$, respectively) was used as an internal standard. The numbers of the free activation enthalpies were derived by using standard equations or by complete line-shape analysis (software: gNMR, Cherwell Scientific) for complex **3**. The two methods yield similar results for $\Delta G^\ddagger$ within $0.1 \text{ kcal mol}^{-1}$.^[13] All structures were fully optimized without any constraints by using B3LYP^[9] and the LANL2DZ basis set augmented with polarization functions (denoted as LANL2DZp).^[10,11] All structures were characterized as minima or transition states by computation of vibrational frequencies (*xyz* files of the calculated reactant and transition states are available in the Supporting Information). Energies are ZPE-corrected on the same level, and the Gaussian 03 suite of programs was used throughout.^[8]

Supporting Information (see footnote on the first page of this article): Line-shape analysis of the *t*Bu signals of **3**, Eyring plot of the data obtained from the VT ^1H NMR of **3**, linear regression for the Eyring plot, description of the ^1H NMR spectrum of **4** in $[\text{D}_6]\text{dmsO}$ and picture of the temperature-independent ^1H signals for the $\text{H}^{\text{N}+}$ proton and the diastereotopic $\text{CH}^{\text{A}}\text{H}^{\text{B}}$ proton, respectively, as well as the *xyz* files of the calculated (RB3LYP/LANL2DZp) $[\text{In}_4(\text{L}^*)_4]$ complexes (reactant and transition states).

Acknowledgments

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- [1] A. Scheurer, A. M. Ako, R. W. Saalfrank, F. W. Heinemann, F. Hampel, K. Petukhov, K. Gieb, M. Stocker, P. Müller, *Chem. Eur. J.* **2010**, *16*, 4784–4792.
- [2] a) R. W. Saalfrank, H. Maid, A. Scheurer, *Angew. Chem.* **2008**, *120*, 8924–8956; *Angew. Chem. Int. Ed.* **2008**, *47*, 8794–8824; b) M. D. Ward, *Chem. Commun.* **2009**, 4487–4499; c) B. Birkmann, R. Frölich, F. E. Hahn, *Chem. Eur. J.* **2009**, *15*, 9325–9329; d) F. Li, J. K. Clegg, P. Jensen, K. Fisher, L. F. Lindoy, G. V. Meehan, B. Moubaraki, K. S. Murray, *Angew. Chem.* **2009**, *121*, 7193–7197; *Angew. Chem. Int. Ed.* **2009**, *48*, 7059–7063; e) P. Mal, D. Schultz, K. Beyeh, K. Rissanen, J. R. Nitschke, *Angew. Chem.* **2008**, *120*, 8421–8425; *Angew. Chem. Int. Ed.* **2008**, *47*, 8297–8301; f) G. Aromí, P. Gamez, J. Reedijk, *Coord. Chem. Rev.* **2008**, *252*, 964–989; g) M. D. Pluth, K. N. Raymond, *Chem. Soc. Rev.* **2007**, *36*, 161–171; h) E. S. Barrett, T. J. Dale, J. Rebek Jr., *J. Am. Chem. Soc.* **2007**, *129*, 8818–8824; i) J. Heo, Y.-M. Jeon, C. A. Mirkin, *J. Am. Chem. Soc.* **2007**, *129*, 7712–7713; j) H.-B. Yang, A. M. Hawkrige, S. D. Huang, N. Das, S. D. Bunge, D. C. Muddiman, P. J. Stang, *J. Am. Chem. Soc.* **2007**, *129*, 2120–2129; k) R. M. McKinlay, J. L. Atwood, *Angew. Chem.* **2007**, *119*, 2446–2449; *Angew. Chem. Int. Ed.* **2007**, *46*, 2394–2397; l) E. G. Bardaji, E. Freisinger, B. Costisella, C. A. Schalley, W. Brüning, M. Sabat, B. Lippert, *Chem. Eur. J.* **2007**, *13*, 6019–6039; m) C. M. Zaleski, J. W. Kampf, T. Mallah, M. L. Kirk, V. L. Pecoraro, *Inorg. Chem.* **2007**, *46*, 1954–1956; n) C. Pariya, C. R. Sparrow, C.-K. Back, G. Sandi, F. R. Fronczek, A. W. Maverick, *Angew. Chem.* **2007**, *119*, 6421–6424; *Angew. Chem. Int. Ed.* **2007**, *46*, 6305–6308; o) M. Albrecht, R. Fröhlich, *Bull. Chem. Soc. Jpn.* **2007**, *80*, 797–808; p) T. Yamaguchi, S. Tashiro, M. Tominaga, M. Kawano, T. Ozeki, M. Fujita, *Chem. Asian J.* **2007**, *2*, 468–476; q) V. Corradini, R. Biagi, U. del Pennino, V. De Renzi, A. Gambardella, M. Alfronte, C. A. Muryn, G. A. Timco, R. E. P. Winpenny, *Inorg. Chem.* **2007**, *46*, 4937–4943.
- [3] For a review of NMR spectroscopy in supramolecular coordination chemistry, see: A. Pastor, E. Martínez-Viviente, *Coord. Chem. Rev.* **2008**, *252*, 2314–2345.
- [4] For examples on VT NMR studies of supramolecular aggregates, e.g. ref.^[2a], see: a) R. W. Saalfrank, B. Demleitner, H. Glaser, H. Maid, D. Bathelt, F. Hampel, W. Bauer, M. Teichert, *Chem. Eur. J.* **2002**, *8*, 2679–2683; b) T. Beissel, R. E. Powers, T. N. Parac, K. N. Raymond, *J. Am. Chem. Soc.* **1999**, *121*, 4200–4206; c) R. L. Paul, S. P. Argent, J. C. Jeffery, L. P. Harding, J. M. Lynam, M. D. Ward, *Dalton Trans.* **2004**, 3453–3458; d) M. Albrecht, I. Janser, H. Houjou, R. Fröhlich, *Chem. Eur. J.* **2004**, *10*, 2839–2850; e) A. Demšar, J. Košmrlj, S. Petriček, *J. Am. Chem. Soc.* **2002**, *124*, 3951–3958; f) R. W. Saalfrank, C. Deutscher, H. Maid, A. M. Ako, S. Sperner, T. Nakajima, W. Bauer, F. Hampel, B. A. Heß, N. J. R. van Eikema Hommes, R. Puchta, F. W. Heinemann, *Chem. Eur. J.* **2004**, *10*, 1899–1905; g) M. Greenwald, D. Wessely, I. Goldberg, Y. Cohen, *New J. Chem.* **1999**, *23*, 337–344; h) M. Fuss, H.-U. Siehl, B. Olenyuk, P. J. Stang, *Organometallics* **1999**, *18*, 758–769.
- [5] R. W. Saalfrank, H. Maid, A. Scheurer, F. W. Heinemann, R. Puchta, W. Bauer, D. Stern, D. Stalke, *Angew. Chem.* **2008**, *120*, 9073–9077; *Angew. Chem. Int. Ed.* **2008**, *47*, 8941–8945.
- [6] Detailed evaluation of the X-ray data of the homochiral crystals of chiral space group $P2_12_12_1$ of **3** indicate that in the crystal **3** is slightly distorted, negligible in solution.
- [7] To the best of our knowledge, this is the first report on the interconversion of a tetrahedral complex with $[M_4L_4]$ stoichiometry (L = face-centered ligand). Analogous processes were observed for $[M_4L_6]$ systems (L = edge-bridging ligand).^[4a,4b]
- [8] M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, J. A. Montgomery Jr., T. Vreven, K. N. Kudin, J. C. Burant, J. M. Millam, S. S. Iyengar, J. Tomasi, V. Barone, B. Mennucci, M. Cossi, G. Scalmani, N. Rega, G. A. Petersson, H. Nakatsuji, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, M. Klene, X. Li, J. E. Knox, H. P. Hratchian, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, P. Y. Ayala, K. Morokuma, G. A. Voth, P. Salvador, J. J. Dannenberg, V. G. Zakrzewski, S. Dapprich, A. D. Daniels, M. C. Strain, O. Farkas, D. K. Malick, A. D. Rabuck, K. Raghavachari, J. B. Foresman, J. V. Ortiz, Q. Cui, A. G. Baboul, S. Clifford, J. Cioslowski, B. B. Stefanov, G. Liu, A. Liashenko, P. Piskorz, I. Komaromi, R. L. Martin, D. J. Fox, T. Keith, M. A. Al-Laham, C. Y. Peng, A. Nanayakkara, M. Challacombe, P. M. W. Gill, B. Johnson, W. Chen, M. W. Wong, C. Gonzalez, J. A. Pople, *Gaussian 03*, Revision C.02, Gaussian, Inc., Wallingford CT, **2004**.
- [9] a) A. D. Becke, *J. Phys. Chem.* **1993**, *97*, 5648–5652; b) C. Lee, W. Yang, R. G. Parr, *Phys. Rev. B* **1988**, *37*, 785–789; c) P. J. Stephens, F. J. Devlin, C. F. Chabalowski, M. J. Frisch, *J. Phys. Chem.* **1994**, *98*, 11623–11627.
- [10] a) P. J. Hay, W. R. Wadt, *J. Chem. Phys.* **1985**, *82*, 270–283; b) P. J. Hay, W. R. Wadt, *J. Chem. Phys.* **1985**, *82*, 284–298; c) P. J. Hay, W. R. Wadt, *J. Chem. Phys.* **1985**, *82*, 299–310; d) S. Huzinaga (Ed.), *Gaussian Basis Sets for Molecular Calculations*, Elsevier, Amsterdam, **1984**.
- [11] The performance of this method for enantiomerization mechanisms is documented, e.g. ref.^[4d], see: a) R. Puchta, R. Meier, N. van Eikema Hommes, R. van Eldik, *Eur. J. Inorg. Chem.* **2006**, 4063–4067; b) R. Puchta, N. van Eikema Hommes, R. Meier, R. van Eldik, *Dalton Trans.* **2006**, 3392–3395.
- [12] The *t*Bu groups in $(L^*)^{3-}$ were replaced by hydrogen atoms $[(L^*)^3]$, to avoid corrupted transition states by rotating methyl groups. The initial coordinates for the DFT calculations were derived from the X-ray structure. In $[In_4(L^*)_4]$ all nitrogen lone pairs point outside.
- [13] a) S. Braun, H.-O. Kalinowski, S. Berger, *100 and More Basic NMR Experiments*, VCH, Weinheim, **1996**; b) I. Sandström, *Dynamic NMR Spectroscopy*, Academic Press, London, **1982**.

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